



香港中文大學化學系  
Department of Chemistry  
THE CHINESE UNIVERSITY OF HONG KONG

# The Chinese University of Hong Kong

## Department of Chemistry Research Seminar Series



Professor Tiow-Gan ONG  
Academic Deputy Director  
Institute of Chemistry,  
Academia Sinica (AS),  
Taiwan

Carbones with its  
Elusive Bonding  
Description and Broad  
Implication  
Complementary to  
NHC-Carbenes

27 Sep 2023 (Wednesday)  
2:30 PM  
LHC 101

Contact Person:  
Professor Fuk Yee KWONG



# Carbones with its Elusive Bonding Description and Broad Implication Complementary to NHC-Carbenes

Wen-Ching Chen,<sup>a</sup> Ting-Hsuan Wang,<sup>a</sup> Tiow-Gan Ong,<sup>a,b\*</sup>

<sup>a</sup>Institute of Chemistry, Academia Sinica, Taipei, Taiwan, ROC. <sup>b</sup>Department of Chemistry, National Taiwan University, Taipei, Taiwan, ROC.

Email: tgong@gate.sinica.edu.tw

Carbones ( $L \rightarrow C \leftarrow L$ ) have emerged recently as a new class of organic molecules featuring carbon(0) directly stabilized by two electron-rich groups (L) through Lewis donor-acceptor interaction.<sup>1</sup> Other mesomeric features can also be understood in terms of allenic or zwitterionic form (see **Figure 1**). Owing to the peculiar bonding situation and the zero-valent nature of the central atoms, carbones have attracted much attention in the chemical community as NHC alternatives because their strong  $\sigma$ -donating ability broadly impacts transition-metal coordination, small molecule activation, main-group chemistry, redox non-innocent coordination, and catalysis.<sup>2</sup> This presentation will describe the synthetic preparation and chemical properties of the carbone as well as its application toward supporting metallic complexes for catalysis in tandem photoredox, cross-coupling reaction via tandem C-H and C-O bond activation and a new spin in diversifying FLP reactivity with co-modulator benzyl alcohol.

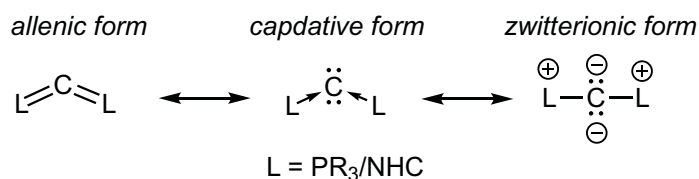


Figure 1. Mesomeric form: bonding situation of carbones.

## References

- [1] R. Tonner, F. Öxler, B. Neumüller, W. Petz, G. Frenking, *Angew. Chem.* **2006**, *118*, 8206–8211.
- [2] a) B. Inés, M. Patil, J. Carreras, R. Goddard, W. Thiel, M. Alcarazo, *Angew. Chem.* **2011**, *123*, 8550–8553; *Angew. Chem. Int. Ed.* **2011**, *50*, 8400–8403; b) C. Esterhuysen, G. Frenking, *Chem.-Eur. J.* **2011**, *17*, 9944–9956; c) D. Martin, M. Soleilhavoup, G. Bertrand, *Chem. Sci.* **2011**, *2*, 389–399; d) W.-C. Chen, Y.-C. Hsu, C.-Y. Lee, G. P. A. Yap, T.-G. Ong, *Organometallics* **2013**, *32*, 2435–2442; e) W.-C. Chen, C.-Y. Lee, B.-C. Lin, Y.-C. Hsu, J.-S. Shen, C.-P. Hsu, G. P. A. Yap, T.-G. Ong, *J. Am. Chem. Soc.* **2014**, *136*, 914–917; f) X. Chen, Z. Li, G. Frenking, I. Fernández, L. Zhao, H. Grützmacher, *Chem.-Eur. J.* **2019**, *25*, 7912–7920; g) M. J. Goldfogel, C. C. Roberts, S. J. Meek, *J. Am. Chem. Soc.* **2014**, *136*, 6227–6230; h) C. C. Roberts, D. M. Matias, M. J. Goldfogel, S. J. Meek, *J. Am. Chem. Soc.* **2015**, *137*, 6488–6491; i) J. S. Marcum, C. C. Roberts, R. S. Manan, T. N. Cervarich, S. J. Meek, *J. Am. Chem. Soc.* **2017**, *139*, 44, 15580–15583; j) J. S. Marcum, T. N. Cervarich, R. S. Manan, C. C. Robert, S. J. Meek, *ACS Catal.* **2019**, *9*, 5881–5889.